

A scenic waterfall cascading over mossy rocks into a pool of water in a lush forest. The water is captured with a long exposure, creating a soft, silky texture. The surrounding area is filled with vibrant green foliage and moss, creating a serene and natural atmosphere. The waterfall is the central focus, with water flowing from the top left towards the bottom center, where it meets a pool of water. The pool is surrounded by large, smooth rocks and is partially shaded by the surrounding forest. The overall scene is peaceful and idyllic, typical of a forest waterfall.

Blood Gas SAQs

Emergency Medicine Fellowship Program

How to use this book:

1) Complete SAQs

- < 3 months until exam: Exam conditions – focus on clear answers ‘to time’
- 3-6 months until exam: Transition towards exam conditions
- > 6 months until exam: Open book is ok, ‘focus on good answers and developing knowledge acquisition

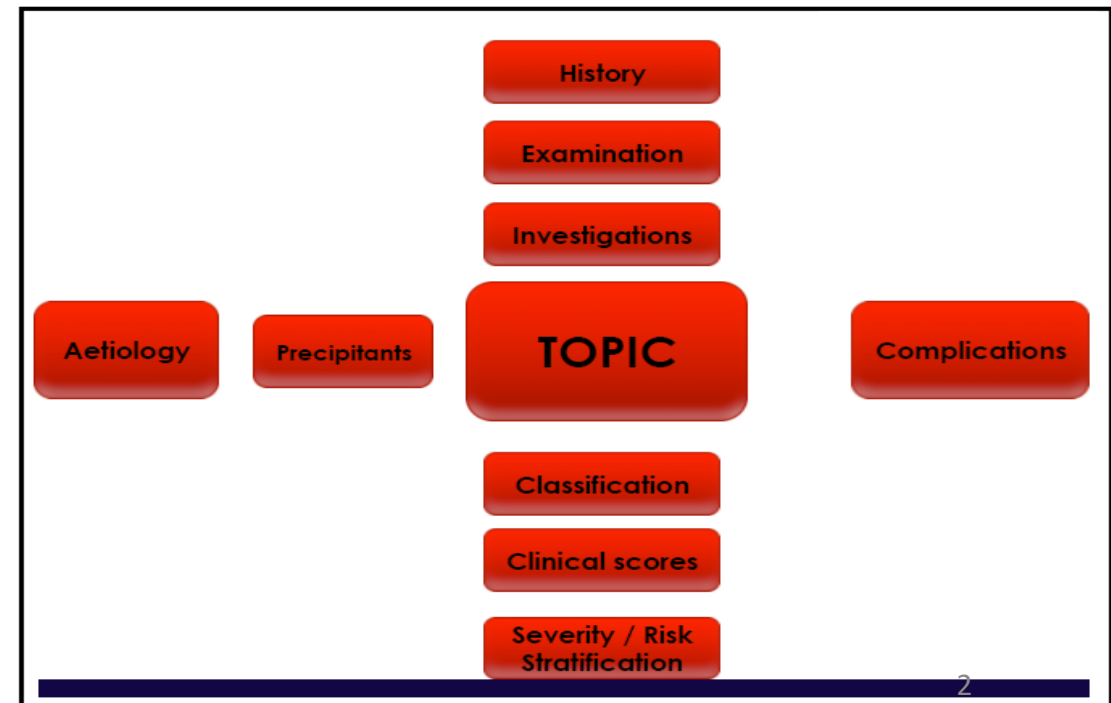
2) Read and study around SAQ

- Use each SAQ as motivation to study around the individual topic
- Think about the different ways the topic could come up in the SAQ exam (use the topic structure provided by APEM course)

3) Write SAQs to further develop this program

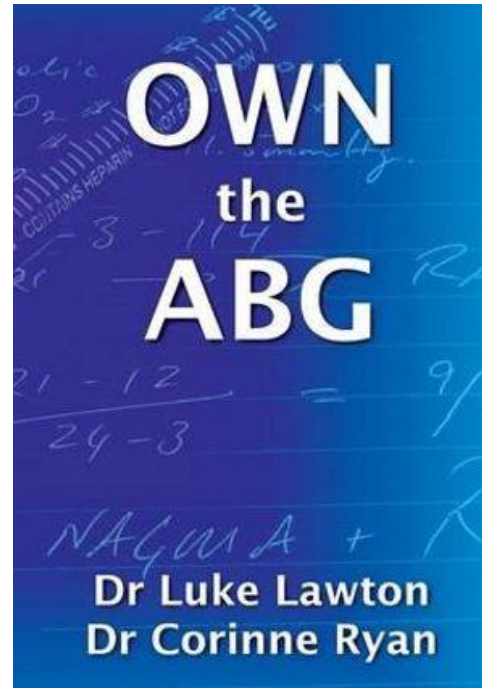
- Review syllabus of medical expertise
- Create SAQs relating to topics not covered in this book, please format to be in exam-format, include answers
- This will help further develop this program as well as help you think like an examiner
- Return to ben.shepherd86@gmail.com

ALL THE BEST!



3.14 Acid Base Disorders

- a) Interpretation of arterial blood gases I Ex
 - i) Alveolar gas equation I H
 - ii) A-a gradient I H
- b) Metabolic acidosis I H
- c) Metabolic alkalosis I H
- d) Respiratory acidosis I H
- e) Respiratory alkalosis I H
- f) Anion gap I Ex
- g) Osmolar gap I H
- h) Indications for the administration of sodium bicarbonate D H



Expectation of Answer of Blood Gas Question

- 1) State primary disturbance
 - Discuss provisional and DDx
- 2) Identify other disturbances
 - Discuss provisional and DDx
- 3) Assess for compensation
- 4) Assess/correct electrolytes
- 5) Assess Oxygenation

Finally

Summarise issues and management priorities

COLUMN "LO" – CATEGORIES OF LEARNING OBJECTIVES

DIS - Diseases/Injuries/Symptoms	D - Pharmacological & toxicological agents
E - Physical Examination	P - Procedures
I - Investigations	Eq - Equipment
M - Medical Interventions	T - Theories
	S - Systems
	NCI - Non-clinical/clinical interface

COLUMN "LP" – LEVELS OF PRACTICE

Ex - Expert
H - High
G - General

DDx Lactaemia

Type A (\downarrow O ₂ delivery)	Type B
Shock Other: -Severe hypoxia -Severe anaemia -CO/CN poisoning	B1 (medical problems) -sepsis -liver failure -thiamine deficiency -diabetes B2 (drugs) -metformin / iron / salicylates and many more B3 (inborn errors)

DDx HAGMA

K: Ketones [Diabetic / Starvation / Alcoholic]

U: Uraemia

L: Lactate

D: Drugs [not directly via lactate]

- toxic alcohols: methanol / ethylene glycol
- salicylates
- toluene

pH 7.4

HCO₃ 24

pCO₂ 40

pO₂ 100

Lactate 2

Delta Ratio = $\frac{AG - 12}{24 - HCO_3}$

< 0.4 = Isolated NAGMA

0.4-0.8 = Mixed NAGMA + HAGMA

1-2 = Isolated HAGMA

>2 = HAGMA + metabolic alkalosis

DDx NAGMA

Hyperchloraemia [e.g. NaCl]

GI loss of HCO₃ [diarrhoea / fistulas]

Renal Loss [Acetazolamide,
Spironolactone, RTA]

Adrenal Insufficiency

Early Uraemia /Recovering DKA etc

Expected HCO₃ (1-2-4-5 rule)

1: Acute Resp Acidosis CO₂ ↑ 10mmHg = HCO₃ ↑ by 1mmol/L

2: Acute Resp Alkalosis CO₂ ↓ 10mmHg = HCO₃ ↓ by 2mmol/L

4: Chronic Resp Acidosis: every CO₂ ↑10mmHg = HCO₃ ↑ by 4mmol/L

5 Chronic Resp Alkalosis CO₂ ↓ 10mmHg = HCO₃ ↓ by 5mmol/L

DDx Metabolic Alkalosis

Chloride Sensitive U Cl < 20	Chloride Resistant U Cl >20
GI Loss [Vomiting / Diarrhoea / NG suction] Diuretic use Exogenous Alkali (Mild Alkali, NaHCO ₃) Contraction Alkalosis (volume depletion)	Mineralocorticoid Excess [Cushings / Conns / Adrenal Hyperplasia]

Aa gradient (normal = 4 + age/4)

A-a gradient = PAO₂ – paO₂

PAO₂ = fiO₂ x 713 – pCO₂/0.8

P/f ratio: paO₂ / fiO₂ ratio

Clinical indicator of hypoxia

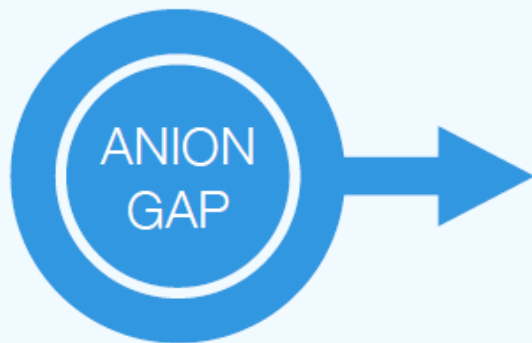
Normal > 500 (at sea level)

Corrected Values

K⁺ for pH: every 0.1 away from pH 7.4, K⁺ changes by 0.5mmol/L in opposite direction

Na⁺ for BSL: Na + (BSL-5) /3

THE 3 ACIDOSIS EXAM RULES



Rule 1

If you see a metabolic acidosis, you **must** calculate the **anion gap**

$$\text{ANION GAP} = [\text{Na}] - [\text{HCO}_3] - [\text{Cl}]$$

Normal 12 (range 6-15)
Albumin correction = $\text{AG} + \frac{1}{4}(44 - \text{albumin})$



Rule 2

If the anion gap is elevated, you should calculate the **delta ratio**

$$\text{DELTA RATIO} = \frac{\uparrow \text{ in AG}}{\downarrow \text{ in } [\text{HCO}_3]}$$

$$\text{DELTA RATIO} = \frac{\text{AG} - 12}{24 - [\text{HCO}_3]}$$

<0.8 = combined HAGMA & NAGMA
1-2 = uncomplicated HAGMA
>2 = pre-existing metabolic alkalosis

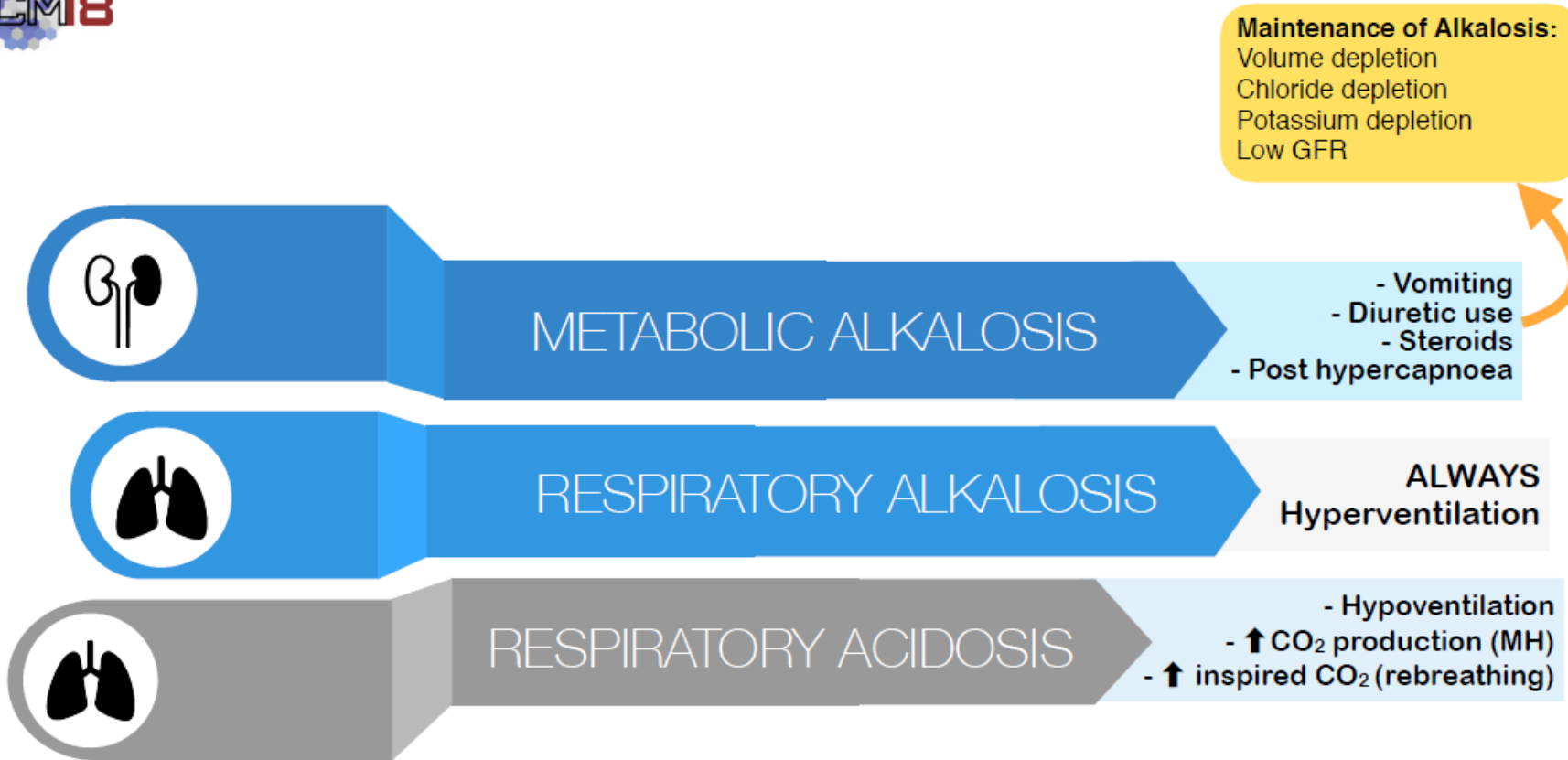


Rule 3

If you see a measured osmolality, you **must** calculate the **osmolar gap**

$$\text{OSMOLAR GAP} = \text{osmolality} - \text{osmolarity}$$

osmolality is *measured*
osmolarity is *calculated*
calc osmolarity = $2[\text{Na}] + \text{urea} + \text{glucose}$



PATTERN RECOGNITION & SPOT DIAGNOSIS

If this is in the stem, then think...

- Urinary pH = RTA
- Polyuria post TBI = mannitol
- 'Young female' = pregnancy
- High glucose = DKA, HHS
- Fluclox/paracetamol with renal/hepatic impairment = pyroglutamic acidosis
- High cholesterol = myxoedema coma
- Osmolality = toxic alcohols

>12

HAGMA

LACTATE	
TOXINS	ethanol, methanol, ethylene glycol, mannitol, salicylates
KETONES	diabetic, alcoholic or starvation ketoacidosis
RENAL	

LTKR: "Left Total Knee Replacement"



$$\text{ANION GAP} = [\text{Na}] - [\text{HCO}_3] - [\text{Cl}]$$

USED CRAP

NAGMA

8-12

Ureterostomy
Small bowel fistula
Extra chloride
Diarrhoea
Carbonic anhydrase inhibitors
Renal tubular acidosis
Addison's disease
Pancreatic duodenal fistula

LAGMA

↓ unmeasured anions	albumin, dilution
↑ unmeasured cations	multiple myeloma, lithium OD, ↑[Ca] or [Mg]
analytical error	↑[Na], viscosity or lipids

<8

OSMOLAR GAP = osmolality - osmolarity

>10

MIME ELK

R A I S E D

Methanol/mannitol
Isopropyl alcohol
Methylene glycol
Ethanol
Ethylene glycol
Lactate
Ketones

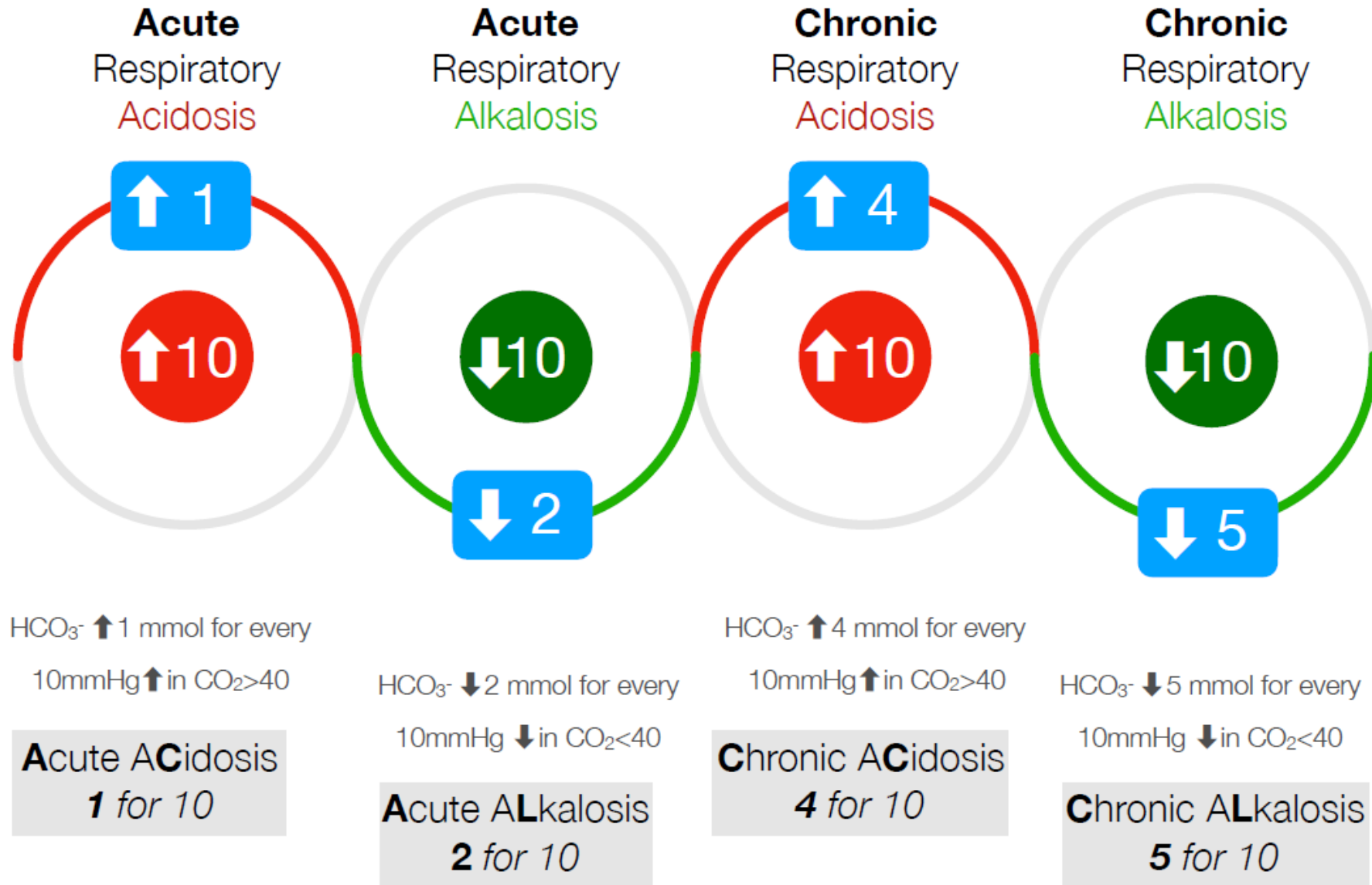
OSMOLAR GAP

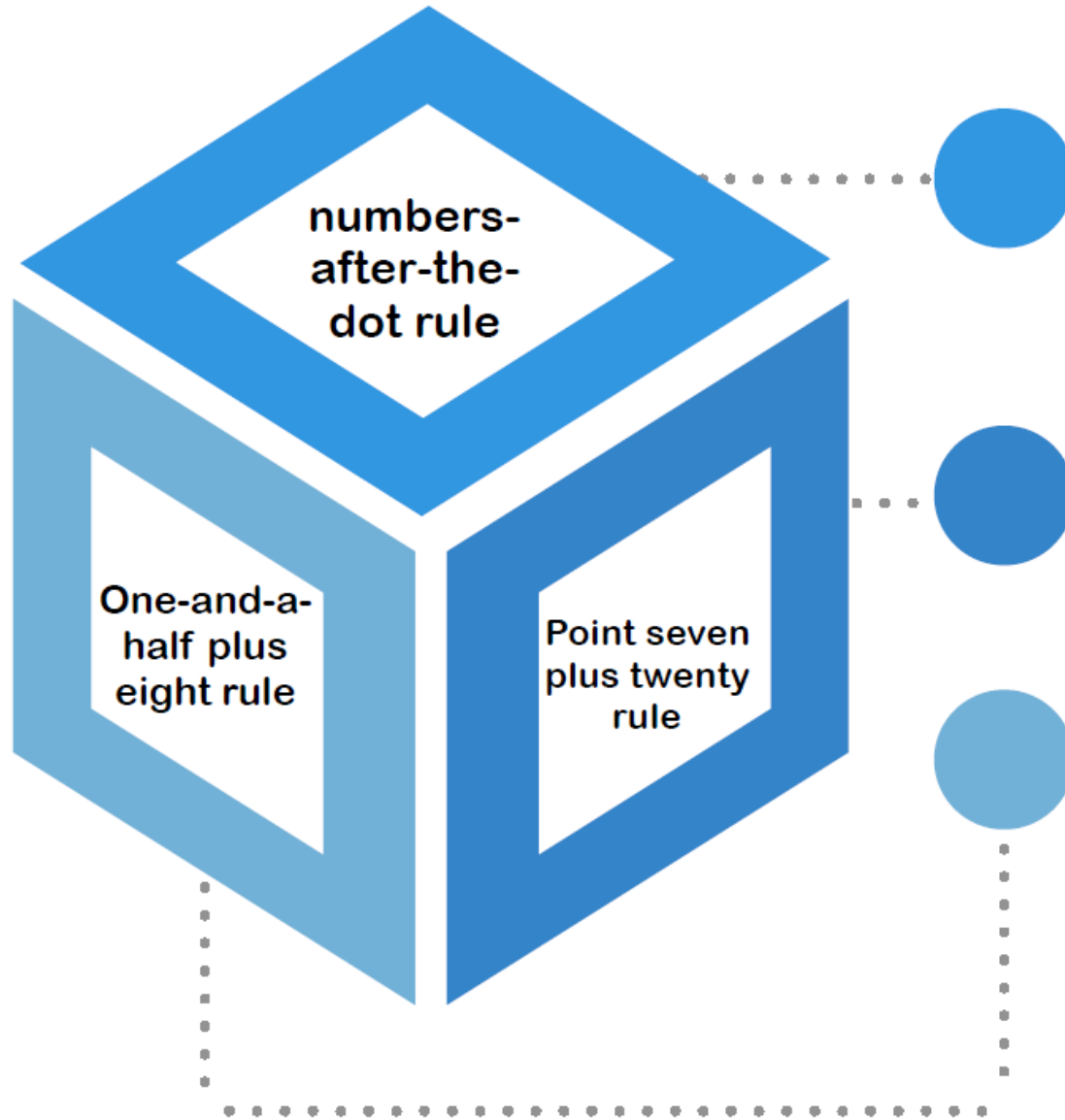
N O R M A L

Pyroglutamic acid
Salicylates

0-10

COMPENSATION RULES





ANY METABOLIC DISORDER

expected $\text{CO}_2 =$ digits after the pH
decimal point

only works where there is a single process

METABOLIC ALKALOSIS

expected $\text{CO}_2 = 0.7 \times [\text{HCO}_3] + 20$

METABOLIC ACIDOSIS

expected $\text{CO}_2 = 1.5 \times [\text{HCO}_3] + 8$

If the pH is **normal**, there must be

- ⊙ 2 or more problems (*mixed picture*)
- ⊙ no problem (*never in the exam*)
- ⊙ pregnant patient (*compensated respiratory alkalosis*)

A-a GRADIENT

$$= PAO_2 - PaO_2$$

= alveolar (calc) - arterial (measured)

Normal A-a
gradient
<15 mmHg

↑ 1-2 mmHg with each decade

↑ 5-7 mmHg for every 10% ↑ FiO_2

Beware the effects of altitude

Hypoventilation does NOT ↑ A-a gradient

Wherever an FiO_2 is given,
you MUST calculate
the A-a gradient

$$PAO_2 = FiO_2(P_B - P_{H_2O}) - (PaCO_2/RQ)$$

Shortcut:

$$PAO_2 \approx FiO_2 \times 500$$

P_B is barometric pressure
 P_{H_2O} is pressure due to water vapour
 RQ is respiratory quotient

Breathing room air at sea level:
 $PAO_2 = 0.21 \times (760 - 47) - (40/0.8)$
 $= 150 - 100$
 $= 100 \text{ mmHg}$

CAUSES OF HYPOXIA with a...

Normal A-a (<15)

- Alveolar hypoventilation
- Low PiO_2
 - $FiO_2 < 0.21$
 - $P_B < 760 \text{ mmHg}$

Raised A-a (≥ 15)

- V/Q mismatch
- R→L shunt (intrapulmonary or cardiac)
- Diffusion defect

A 20 year old male presents with a 3 day history of lethargy and generalised malaise. He is confused and looks unwell. The following venous blood tests are obtained.

pH	7.08		
pCO ₂	20	mmHg	(35-45)
HCO ₃	16	mmol/L	(22-28)
BE	-22	mmol/L	(-3-3)
Lactate	9	mmol/L	(<2.2)
Glucose	44	mmol/L	(3.9-5.8)
K	2.7	mmol/L	(3.5-4.2)
Na	160	mmol/L	(135-145)
Cl	124	mmol/L	(95-110)
Creatinine	70	micromol/L	(60-110)
Urea	5.3	mmol/L	(3-8)

i. Interpret the above results, and show any relevant calculations (5 marks)

ii. Outline your immediate management priorities (5 marks)

A 2 year old boy is brought to your emergency department by his grandparents having found him unresponsive at their home. He has no significant past medical history.

On examination he is very drowsy with a GCS of 7 (E2, V1, M4) and has the following observations:

HR	140	/min
BP	80/50	mmHg
RR	18	/min
O2 saturation	100%	on 2L/min O2 via nasal cannulae
Temperature	36.5	oC

His initial blood results are shown:

Venous blood gas

			Reference Range
pH	7.1		7.35 – 7.45
pO2	50	mmHg	
pCO2	37	mmHg	40 – 52
HCO3-	11	mmol/L	24 – 32
Lactate	8.8	mmol/L	0.5 – 2.0
Na+	143	mmol/L	135 – 145
K+	3.8	mmol/L	3.5 – 4.8
Cl-	110	mmol/L	95 – 110
Glucose	1.8	mmol/L	3.0 – 6.0

1. List five (5) significant abnormalities on the blood gas

2. What is the acid/base disturbance? Show your calculations.

3. List five (5) relevant differential diagnoses.

A 20 year old woman with a history of diabetes mellitus presents to your emergency department with fever, acute respiratory distress and chest pain.

Her observations on arrival (pre treatment) are:

GCS	14	(E4, V4, M6)
HR	128	beats/min
BP	85/45	mmHg
SaO ₂	90%	on 10L O ₂ via Hudson mask

Arterial blood gases and serum biochemistry are taken after 20 minutes of supplemental oxygen therapy. Results are given below.

FiO ₂	100%		
pH	7.20	mmHg	(7.36 - 7.44)
pCO ₂	28	mmHg	(35 - 45)
pO ₂	150	mmHg	(85 - 110)
HCO ₃	15	mmol/L	(21 - 28)
Na ⁺	140	mmol/L	(135 - 145)
K ⁺	6.0	mmol/L	(3.2 - 4.3)
Cl ⁻	95	mmol/L	(99 - 109)

i. Calculate this patients alveolar pO₂ (pAO₂), using the alveolar gas equation. Show your calculations and results.

ii. Using your answer from question 1, calculate this patients alveolar-arterial (A-a gradient) for oxygen.

iii. List two (2) possible causes of your findings and justify your answer.

	Cause	Justification
1		
2		

iv. Calculate the anion gap from these results.

v. List two (2) likely causes of the anion gap finding.

1 _____

2 _____

vi. Calculate the predicted serum K⁺ if the patient's pH was corrected to 7.40, assuming no other factors affect it.

A 32 year-old male smoker is B1B to the Emergency Department of a tertiary hospital following a fire at his metal cleaning and reclaiming shop. He has a fluctuating level of consciousness. His current examination shows :

GCS	13	(E4 M5 V4)
Temperature	36.5	degrees Celsius
RR	10	/min
BP	88/55	mmHg
HR	130	/min
O2 Saturation	96%	on O ₂ 6 L/minute via face mask

He has no evidence of airway compromise, burns or other significant injury.

The venous blood gas is taken and is available in the props booklet on page 5

- i. List three (3) important abnormalities. State the significance of each abnormality (6 marks)

	Abnormality (3 marks)	Significance (3 marks)
1		
2		
3		

PH	7.18		(7.35 - 7.45)
PO ₂	100	mmHg	
PCO ₂	45	mmHg	(40 - 52)
HCO ₃ ⁻	12	mmol/L	(24 - 32)
COHb	18	%	
Na ⁺	145	mmol/L	(135 - 145)
K ⁺	4.8	mmol/L	(3.5 - 4.8)
Cl ⁻	107	mmol/L	(95 - 110)
Glucose	10.8	mmol/L	(3.0 - 6.0)
Lactate	16	mmol/L	(0.5 - 2.0)

- ii. Provide two (2) calculations that will assist your diagnostic process. State how each calculation will contribute to the diagnosis (4 marks)

Calculation (2 marks)	Contribution to diagnosis (2 marks)

- iii. List the two (2) most clinically important differential diagnoses. (2 marks)

1 _____

2 _____

- iv. List two (2) specific treatments that you would commence. Provide details for each treatment including doses and routes where applicable. (4 marks)

	Specific treatment (2 marks)	Details (2 marks)
1		
2		

A previously well 55 year old woman presents with a complaint of severe vomiting for 5 days. Her observations are as follows:

HR	110	/min
BP	100/60	mmHg
Temp	36.8	oC

A venous blood gas is performed and is shown below:

pH	7.23		(7.35 - 7.45)
pO2	40	mmHg	
pCO2	22	mmHg	(40 - 52)
HCO3	9	mmol/L	(24 - 32)
Na	140	mmol/L	(135 - 145)
K	3.4	mmol/L	(3.5 - 4.8)
Cl	77	mmol/L	(95 - 110)

i. What is the primary acid/base disturbance? (1 mark)

ii. Calculate the anion gap - show your calculation. (2 marks)

iii. Is there adequate compensation? - show your calculation. (2 marks)

iv. What is the delta ratio? - show your calculation. (2 marks)

v. What does this delta ratio result tell us in this clinical setting? (2 marks)

vi. Provide a unifying explanation for these gas results. (2 marks)

A 75 year old male is brought to your ED via ambulance, with Respiratory Distress, fever and confusion. On assessment the patient is confused and combative. His vital signs on arrival are:

Temp	38.7	°C
HR	120	bpm, irregular
BP	110/60	mm Hg
RR	32	/min
O2 sats	91%	on High flow Nasal prong O2

The patients arterial blood gas is reprinted below

PH	7.21		(7.32 - 7.42)
PCO2	64	mmHg	(41 - 51)
PO2	75	mmHg	
HCO3	21	mmol/L	(20 - 40)
BE	-6	mmol/L	(-3.0 - +3.0)
Na	148	mmol/L	(135-145)
Cl	105	mmol/L	(95 - 107)

a) Assuming the patients K is normally 4.0, based on these acid base abnormalities Calculate the following Parameters showing your calculations. (2 marks)

Expected potassium (K)	Delta Ratio

b) What is the acid base abnormality on these results? Justify your answer (2 marks)

c) List Three (3) key differential diagnosis in this case; For each differential list Two (2) investigations you would perform, with a justification for each (15 marks)

	DDx	Investigation	Justification
1		a.	
		b.	
2		a.	
		b.	
3		a.	
		b.	

A 48 year old woman with shortness of breath for 2 days. She has a history of diabetes and depression. Her LMO has been treating her for an URTI without improvement. On assessment she is awake and oriented. Her vital signs are:

BP	150/90 mmHg
HR	120 /min
Sats	99% RA
Temp	36.8°C

Her arterial blood gas is shown in PROPS BOOKLET ; Page 12 .

a) What is the acid base abnormality? (1 mark)

b) Complete the table below showing requested calculations (4 marks)

	Calculations and results
A-a gradient	
Delta gap	
Expected K (Normal = 4)	
Corrected Na	

PH 7.10 (7.32 - 7.42)

PCO2 39 mmHg (41 - 51)

PO2 94 mmHg

HCO3 12 mmol/L (20 - 40)

BE -16 mmol/L (-3.0 - +3.0)

Na 128 mmol/L (135 - 145)

K 5.2 mmol/L (3 - 5)

Cl 97 mmol/L (95 - 107)

Glu 47 mmol/L (3.0 - 10.0)

Cr 125 mmol/L (50 - 120)

c) List three (3) essential treatments for this patient including dosage. (6 marks)

	Treatment	Dosage
1		
2		
3		

Four hours after commencement of treatment the patient becomes drowsy and responsive to only painful stimuli.

d) List three (3) essential investigations that should urgently be performed with one justification for each. (6 marks)

	Investigation	Justification
1		
2		
3		

A 30 year old man presents to the emergency department with new onset bilateral lower leg weakness. He denies any past medical history. His venous blood gas is reproduced here.

pH	7.399	(7.320-7.420)
pCO ₂	32.8 mmHg	(41.0-51.0)
pO ₂	56.0 mmHg	(20-40)
HCO ₃ ⁻	19.9 mmol/L	(21.0-30.0)
Base Excess	-4.1 mmol/L	(-3.0-3.0)
Na ⁺	143 mmol/L	(135-145)
K ⁺	1.5 mmol/L	(3.5-5.0)
Cl ⁻	113 mmol/L	(95-107)
Glucose	8.3 mmol/L	(3.0-10.0)
Lactate	2.3 mmol/L	
Creatinine	51 µmol/L	(50-120)

i. List three (3) abnormalities and explain their significance.

	Abnormalities	Significance
1		
2		
3		

ii. Calculate two (2) values and state how they contribute to your diagnosis.

1. _____

2. _____

_____ / 4

iii. List four (4) differential diagnoses for this patient's presentation.

1. _____

2. _____

3. _____

4. _____

_____ / 4

iv. List three (3) changes on an ECG that could be expected for this patient based on his blood results.

1. _____

2. _____

3. _____

An 84 year old man is brought to your emergency department following a high speed car accident. He has signs of multiple left rib fractures. Two hours after arriving in the emergency department he becomes more breathless and distressed.

His observations are:

- GCS 14
- HR 75 bpm
- BP 100/60
- RR 24

An arterial blood gas is performed

			Reference Range
pH	7.14		(7.35-7.45)
pCO ₂	60	mmHg	(35-45)
pO ₂	114		
HCO ₃ ⁻	17	mmol/L	(21-28)
Lactate	1.4	mmol/L	(< 2.0)
FiO ₂	50	%	
Na ⁺	139	mmol/L	(135-145)
K ⁺	4.8	mmol/L	(3.2-4.3)
Cl ⁻	116	mmol/L	(99-109)
Glucose	11.3	mmol/L	(3.0-6.0)

a. Calculate the patient's A-a gradient and show the formula/s used in the calculation (3 Marks)

b. Calculate the patient's expected pCO₂ and show the formula/s used (2 Marks)

c. Calculate the patient's expected HCO₃⁻ increase assuming all changes are acute show the formula/s used (2 Marks)

d. List 6 potential causes of the patient's ABG results (3 Marks)

1.

2.

3.

4.

5.

6.

A 23 year old male with a decreased level of consciousness is being assessed in your ED. His arterial blood gas results with reference ranges are:

			Reference Range
FIO ₂	0.3		
pH	6.9		(7.35-7.45)
pCO ₂	10	mmHg	(37-45)
pO ₂	147	mmHg	(80-95)
Bicarbonate	2	mmol/L	(22-28)
Base excess	-30		(-3 - +3)
O ₂ saturation	98	%	(> 95)
Lactate	7.1	mmol/L	(< 1.3)
Na ⁺	140	mmol/L	(134-146)
K ⁺	6.0	mmol/L	(3.4-5.0)
Cl ⁻	105	mmol/L	(98-106)
Creatinine	0.1	mmol/L	(0.06-0.12)
Urea	4.8	mmol/L	(3.0-8.0)
Glucose	5.2	mmol/L	(3.5-5.5)
Osmolality	360	mOsm/L	(275-295)

a. List 4 key abnormalities on this patient's gas (2 Marks)

1. _____
2. _____
3. _____
4. _____

b. Calculate the patient's anion gap and write the formula/s used (2 Marks)

c. Calculate the patient's osmolar gap and write the formula/s used (3 Marks)

d. List 6 causes for a raised osmolar gap (3 Marks)

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____

A 55 year old Female with PMH of CCF and DM, is brought to your emergency department by ambulance with vomiting and drowsiness. here is no clinical signs to suggest trauma. Her Vital signs are :

BP	95/60	mmHg
HR	110	bpm
RR	28	bpm
O2Sat	94%	RA
T	37	deg celsius
GCS	14	(E3V5M6)

Her venous blood gas is shown in PROPS BOOKLET; PAGE 6.

a) What is the likely explanation for these gas results in this patient (2 marks)

1. _____
2. _____
3. _____
4. _____
5. _____

b) Is anion gap normal or abnormal? show your calculation. (2 mark)

Venous blood gas

PH	7.41		(7.32 - 7.42)
PCO2	32	mmHg	(41 - 51)
HCO3	19	mmol/L	
Na	144	mmol/L	(135 - 145)
K	2.7	mmol/L	(3 - 5)
Cl	90	mmol/L	(95 - 107)
Glu	25	mmol/L	(3.0 - 10.0)
Lactate	1.8	mmol/L	(0.5 - 2)

c) Calculate the Approximate Delta Gap and show your calculation. (2 marks)

d) What acid base disturbances does the Delta gap suggest? (1 mark)

e) List Four (4) expected changes with the ECG for Hypokalaemia.(4 marks)

1.

2.

3.

4.

A 32 year-old male smoker is BIBA to the Emergency Department of a tertiary hospital following a fire at his metal cleaning and reclaiming shop. He has a fluctuating level of consciousness. His current examination shows:

GCS	13	(E4 M5 V4)
BP	88/55	mmHg
HR	130	bpm
Temperature	36.5	deg Celsius
RR	10	bpm
O2 Saturation	96%	on O2 6 L/minute via face mask

He has no evidence of airway compromise, burns or other significant injury.
The venous blood gas is taken and is available in **PROPS BOOKLET; PAGE 13**.

a) List three (3) important abnormalities. State One significance for each abnormality (6 marks)

1. _____

2. _____

3. _____

Venous blood gas

PH 7.18 (7.32 - 7.42)

PCO2 45 mmHg (41 - 51)

PO2 90 mmHg

COHb 18 %

HCO3 12 mmol/L (24 - 32)

Na 145 mmol/L (135 - 145)

K 4.8 mmol/L (3 - 5)

Cl 107 mmol/L (95 - 107)

Glu 10.8 mmol/L (3 - 10)

Lactate 16 mmol/L (0.5 - 2)

b) Provide two (2) calculations that will assist your diagnostic process. State how each calculation will contribute to the diagnosis (4 marks)

1. _____

2. _____

c) List the two (2) most clinically important differential diagnoses. (2 marks)

1. _____

2. _____

A 62-year-old male with a history of chronic pancreatitis presents to the Emergency Department with several days of nausea and vomiting.

His biochemistry profile is attached:

Parameter	Patient Value	Normal Adult Range
Arterial Blood Gas		
FiO ₂	0.4	
pH	7.62*	7.35 – 7.45
PCO ₂	62 mmHg* (8.2 kPa)*	36 – 45 (4.6 – 6.0)
PO ₂	133 mmHg (17.5 kPa)	
Bicarbonate	65 mmol/L*	21 – 28
Base Excess	> 30 mmol/L*	-3 – +3
Sodium	149 mmol/L*	135 – 145
Potassium	3.3 mmol/L*	3.5 – 5.2
Chloride	53 mmol/L*	95 – 110
Calcium ionised	0.74 mmol/L*	1.12 – 1.32
Lactate	2.7 mmol/L*	< 1.3
Venous biochemistry		
Urea	34.9 mmol/L*	3.0 – 8.0
Creatinine	431 micromol/L*	60 – 110

1. Interpret the abnormalities in the results. (3 marks)

2. Give likely underlying causes for the main metabolic acid-base disturbances. (3 marks)

3. Describe the clinical features of severe hypocalcaemia. (2 marks)

4. Outline your management of symptomatic hypocalcaemia. (2 marks)

A 32-year-old Aboriginal woman presents to the ED with lethargy. She is previously well, and has not been seen by medical services for the previous 10 years.

Her VBG is attached:

pH	7.25
pCO ₂	24.4
HCO ₃	11.7
Hb	81
Na	140
K	5.1
Ca ⁺⁺	0.81
Cl	98
Gluc	9.6
Lact	0.9

1. What is the most likely cause for the abnormalities on the VBG?
Describe and interpret the abnormalities. (5 marks)

Her creatinine is 1842 with a urea of 62.

2. List 4 possible causes for these findings in a 32-year-old Aboriginal woman, with the most common cause first. (3 marks)

(1) _____

(2) _____

(3) _____

(4) _____

She becomes agitated and appears to be getting ready to leave the ED.

3. What would be some strategies to stop her leaving? (4 marks)

An 89 year old lady presents with generalised weakness, nausea and diarrhoea. She had a fall 3 days earlier and had broken her humerus. She was discharged for orthopaedic follow up as an outpatient. She appears dehydrated, BP 140/80, pulse 80 regular, and is afebrile.

Her venous blood gas is below :

Venous Blood pH POCT	7.073
Venous Blood pO ₂ POCT	23.8 mmHg
Venous Blood pCO ₂ POCT	48.2 mmHg
Venous Blood O ₂ Saturation POCT	33.2 %
Venous Blood HCO ₃ POCT	13.4 mmol/L
Venous Blood Base Excess POCT	-14.8 mmol/L
Venous Blood Oxyhaemoglobin POCT	32.8 %
Venous Blood Inspired Oxygen POCT	21 %
Venous Blood Haemoglobin POCT	97 g/L
Venous Blood Reduced Haemoglobin POCT	66.0 %
Venous Blood Methaemoglobin POCT	0.4 %
Venous Blood Carboxyhaemoglobin POCT	0.8 %
Venous Blood Creatinine POCT	179 umol/L
Venous Blood Sodium POCT	126 mmol/L
Venous Blood Potassium POCT	3.2 mmol/L
Venous Blood Chloride POCT	100 mmol/L
Venous Blood Calcium Ionised POCT	1.25 mmol/L
Venous Blood Glucose POCT	5.6 mmol/L
Venous Blood Lactate POCT	1.1 mmol/L

1. Interpret the blood gas results - what are the likely causes of her acidosis? (5 marks).

2. List 5 further investigations you feel are indicated. Justify your responses (5 marks).

- i.

- ii.

- iii.

- iv.

- v.

A 30 year old man presents to your ED with a 24 hour history of increasing weakness.

An ABG and electrolytes are performed.

ABG

FiO2	21%	
pH	7.26 mmHg	(7.35-7.45)
pCO2	29 mmHg	(35-48)
pO2	101 mmHg	(83-108)
HCO3	13 mmol/L	(21-28)
BE	-13 mmol/L	(-1.5-3.0)

ELECTROLYTES

Na	137 mmol/L	(136-146)
K	1.8 mmol/L	(3.9-5.2)
CL	116 mmol/L	(95-110)
Urea	7.8 mmol/L	(3.1-8.1)
Creat	86 mmol/L	(60-110)
Glc	4.0 mmol/L	(3.9-5.8)
CK	1975 U/L	(29-168)
Trop T	20 ng/L	(<30)

i. Describe and interpret his blood results (8 marks)

ii. List your 3 immediate management priorities (3 marks)

iii. An internal jugular central venous catheter is placed to facilitate potassium replacement. List 4 early and 4 late complications of internal jugular central venous catheter placement (8 marks)

An 84 year old man is brought to your emergency department following a high speed car accident. He has signs of multiple left rib fractures. Two hours after arriving in the emergency department he becomes more breathless and distressed.

His observations are:

- GCS 14
- HR 75 bpm
- BP 100/60
- RR 24

An arterial blood gas is performed

			Reference Range
pH	7.14		(7.35-7.45)
pCO ₂	60	mmHg	(35-45)
pO ₂	114		
HCO ₃ ⁻	17	mmol/L	(21-28)
Lactate	1.4	mmol/L	(< 2.0)
FiO ₂	50	%	
Na ⁺	139	mmol/L	(135-145)
K ⁺	4.8	mmol/L	(3.2-4.3)
Cl ⁻	116	mmol/L	(99-109)
Glucose	11.3	mmol/L	(3.0-6.0)

a. Calculate the patient's A-a gradient and show the formula/s used in the calculation (3 Marks)

b. Calculate the patient's expected pCO₂ and show the formula/s used (2 Marks)

c. Calculate the patient's expected HCO₃⁻ increase assuming all changes are acute show the formula/s used (2 Marks)

d. List 6 potential causes of the patient's ABG results (3 Marks)

1.

2.

3.

4.

5.

6.

A 23 year old male with a decreased level of consciousness is being assessed in your ED. His arterial blood gas results with reference ranges are:

			Reference Range
FIO ₂	0.3		
pH	6.9		(7.35-7.45)
pCO ₂	10	mmHg	(37-45)
pO ₂	147	mmHg	(80-95)
Bicarbonate	2	mmol/L	(22-28)
Base excess	-30		(-3 - +3)
O ₂ saturation	98	%	(> 95)
Lactate	7.1	mmol/L	(< 1.3)
Na ⁺	140	mmol/L	(134-146)
K ⁺	6.0	mmol/L	(3.4-5.0)
Cl ⁻	105	mmol/L	(98-106)
Creatinine	0.1	mmol/L	(0.06-0.12)
Urea	4.8	mmol/L	(3.0-8.0)
Glucose	5.2	mmol/L	(3.5-5.5)
Osmolality	360	mOsm/L	(275-295)

a. List 4 key abnormalities on this patient's gas (2 Marks)

1. _____
2. _____
3. _____
4. _____

b. Calculate the patient's anion gap and write the formula/s used (2 Marks)

c. Calculate the patient's osmolar gap and write the formula/s used (3 Marks)

d. List 6 causes for a raised osmolar gap (3 Marks)

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____